

- 1.** $\text{Fe}_2\text{O}_3 + \text{Al} \xrightarrow{\text{t}^\circ} \text{Al}_2\text{O}_3 + \text{Fe}$.
- 2.** $\text{CaCO}_3 + \text{HCl} \rightarrow \text{CaCl}_2 + \text{CO}_2 + \text{H}_2\text{O}$.
- 3.** $\text{AlCl}_3 + \text{AgNO}_3 \rightarrow \text{Al}(\text{NO}_3)_3 + \text{AgCl}$.
- 4.** $\text{Cu} + \text{FeCl}_3 \rightarrow \text{CuCl}_2 + \text{FeCl}_2$.
- 5.** $\text{Fe} + \text{Cl}_2 \rightarrow \text{FeCl}_3$.
- 6.** $\text{Fe} + \text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2$.
- 7.** $\text{Ca} + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2 + \text{H}_2$.
- 8.** $\text{Fe} + \text{O}_2 \xrightarrow{\text{t}^\circ} \text{Fe}_3\text{O}_4$.
- 9.** $\text{C} + \text{CuO} \xrightarrow{\text{t}^\circ} \text{Cu} + \text{CO}_2$.
- 10.** $\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$.
- 11.** $\text{NaHCO}_3 \xrightarrow{\text{t}^\circ} \text{Na}_2\text{CO}_3 + \text{CO}_2\uparrow + \text{H}_2\text{O}$.
- 12.** $\text{NaHCO}_3 + \text{KOH} \rightarrow \text{Na}_2\text{CO}_3 + \text{K}_2\text{CO}_3 + \text{H}_2\text{O}$.
- 13.** $\text{Na} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\uparrow$.
- 14.** $\text{Mg} + \text{O}_2 \xrightarrow{\text{t}^\circ} \text{MgO}$.
- 15.** $\text{Cu}(\text{OH})_2 \xrightarrow{\text{t}^\circ} \text{CuO} + \text{H}_2\text{O}$.
- 16.** $\text{KNO}_3 \xrightarrow{\text{t}^\circ} \text{KNO}_2 + \text{O}_2$.
- 17.** $\text{CaCO}_3 \xrightarrow{\text{t}^\circ} \text{CaO} + \text{CO}_2$.
- 18.** $\text{Al} + \text{HCl} \rightarrow \text{AlCl}_3 + \text{H}_2$.
- 19.** $\text{Fe}(\text{OH})_2 + 2\text{HNO}_3 \rightarrow \text{Fe}(\text{NO}_3)_2 + 2\text{H}_2\text{O}$.
- 20.** $\text{KClO}_3 \xrightarrow{\text{t}^\circ} \text{KCl} + \text{O}_2$.
- 21.** $\text{NaOH} + \text{Cl}_2 \rightarrow \text{NaCl} + \text{NaClO} + \text{H}_2\text{O}$.
- 22.** $\text{Fe}(\text{OH})_2 + \text{O}_2 \xrightarrow{\text{t}^\circ} \text{Fe}_2\text{O}_3 + \text{H}_2\text{O}$.
- 23.** $\text{Ca} + \text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2$.
- 24.** $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$.
- 25.** $\text{Fe} + \text{Cl}_2 \xrightarrow{\text{t}^\circ} \text{FeCl}_3$.
- 26.** $\text{Fe} + \text{HCl} \rightarrow \text{FeCl}_2 + \text{H}_2$.
- 27.** $\text{Mg}(\text{HCO}_3)_2 + 2\text{Ca}(\text{OH})_2 \rightarrow \text{Mg}(\text{OH})_2 + \text{CaCO}_3 + \text{H}_2\text{O}$.
- 28.** $\text{Ca}(\text{OH})_2 + \text{NaHCO}_3 \rightarrow \text{CaCO}_3 + \text{NaOH} + \text{H}_2\text{O}$.
- 29.** $\text{Ca}(\text{OH})_2 + \text{NH}_4\text{Cl} \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{NH}_3$.
- 30.** $\text{Al} + \text{CuSO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + \text{Cu}$.
- 31.** $\text{Al} + \text{Fe}_3\text{O}_4 \xrightarrow{\text{t}^\circ} \text{Al}_2\text{O}_3 + \text{Fe}$.
- 32.** $\text{Al}_2\text{O}_3 \xrightarrow{\text{dpmc}} \text{Al} + \text{O}_2$.
- 33.** $\text{Al} + \text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + \text{H}_2$.
- 34.** $\text{FeS}_2 + \text{O}_2 \xrightarrow{\text{t}^\circ} \text{SO}_2 + \text{Fe}_2\text{O}_3$.
- 35.** $\text{K}_2\text{SO}_3 + \text{H}_2\text{SO}_4 \xrightarrow{\text{t}^\circ} \text{K}_2\text{SO}_4 + \text{SO}_2 + \text{H}_2\text{O}$.
- 36.** $\text{CO} + \text{Fe}_2\text{O}_3 \xrightarrow{\text{t}^\circ} \text{Fe} + \text{CO}_2$.
- 37.** $\text{CO} + \text{Fe}_3\text{O}_4 \xrightarrow{\text{t}^\circ} \text{FeO} + \text{CO}$.
- 38.** $\text{H}_2\text{SO}_4 + \text{Na} \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2$.
- 39.** $\text{H}_2\text{SO}_4 + \text{Na}_2\text{CO}_3 \rightarrow \text{Na}_2\text{SO}_4 + \text{CO}_2 + \text{H}_2\text{O}$.
- 40.** $\text{H}_2\text{SO}_4 + \text{Ca} \rightarrow \text{CaSO}_4 + \text{H}_2$.
- 41.** $\text{H}_2\text{S} + \text{Pb}(\text{NO}_3)_2 \rightarrow \text{PbS}\downarrow + \text{HNO}_3$.
- 42.** $\text{Na}_2\text{S} + \text{Pb}(\text{NO}_3)_2 \rightarrow \text{PbS}\downarrow + \text{NaNO}_3$.
- 43.** $\text{O}_2 + \text{H}_2\text{S} \rightarrow \text{H}_2\text{O} + \text{SO}_2$.

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